



Quantitative Chemical Analysis

By Herbert C Scurlock

Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1915 Excerpt: .at 130 to constant weight, and calculate the chloride as hydrogen chloride. EXERCISE XIV. Determination of Pico centshoric Acid. (a) By Precipitation as Phosphomolybdate.--Prepare a concentrated solution of ammonium nitrate. Prepare also a molybdate solution as follows: Dissolve 37.5 gms. of ammonium molybdate in 250 cc. of water to which a little ammonia is added. Filter if turbid. Make a mixture of 12.5 cc. each of concentrated nitric acid and water, and while constantly stirring, pour the solution of ammonium molybdate into it. Set the solution aside in a warm place for a few days and then filter before use. Heat the phosphate solution given you nearly to boiling and, while constantly stirring, add 50 cc. of molybdate solution. Stir for a while, maintaining the temperature on a water bath for about ten minutes. Allow the mixture to stand for an hour; then, filter through a...



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